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Electronegativity greater than 1.7 and is a transfer of electrons.
The Lower Electronegativity transfers to the higher electronegativity) t. Ionic Bond. Ions pack in a crystal lattice structure. Lattice Energy.

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the intermolecular attractions resulting from the constant motion of electrons and the creation of instantaneous dipoles; forces act

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between all atoms AND molecules; forces are NOT present in intramolecular forces; increases with atomic mass or molar mass.

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Modern Chemistry Chapter 6 Test

Modern Chemistry • CHAPTER 6 HOMEWORK 6-3 (pp. 167-169) VOCABULARY Write true or false for each statement. 1. The bond length is the distance between two bonded atoms at their maximum potential energy. ____ 2. The energy required to form a chemical bond between two neutral atoms is called bond energy. ____ 3.

HOMEWORK 6-1

Chemistry: Chapter 6 Chemical Bonding Section 1. STUDY. Flashcards. Learn. Write. Spell. Test. PLAY. Match. Gravity. Created by. StorytellerAngel. vocabulary for the Holt Modern Chemistry textbook. Terms in this set (6) chemical bond. a

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mutual electrical attraction between the nuclei and valence electrons of different atoms that binds the atoms ...

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REVIEW Chemical Bonding MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. a.

6 Chemical Bonding - Effingham County Schools / Overview

Modern Chemistry 105 Chapter Test Name Class Date Chapter Test A, continued Use this figure to answer questions 7 and 8. _____ 7. A solution containing 35 g of Li_2SO_4 dissolved in 100 g of water is heated from 10°C to 90°C . According to information in the figure, this temperature change would result in a. an additional 5 g of Li_2SO_4 in ...

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Holt Modern Chemistry Review CHAPTER 6: CHEMICAL BONDING The following pages contain the bulk (but not all) of the information for the chapter 6 test. Focus on this content, but make sure to review class notes, activities, handouts, questions, etc. If you study this document and NOTHING else, you should at least be able to PASS the test.

CHEMISTRY CHAPTER 6 REVIEW.docx - Holt Modern Chemistry ...

Modern Chemistry 37 . Name Section Quiz, continued Class Date 6. Metals are malleable because when struck, one plane of metal atoms can slide past another plane without breaking bonds. b. cannot easily move out of the way. c. moves in a way that maximizes the repulsive forces within the

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Home - David Brearley High School

Modern Chemistry 33 Chapter Test Name Class Date Chapter Test B, continued 15. The energy state of an atom is called its ground state. 16. The number of waves that pass a point in one second is called. 17. When an electron drops from a higher-energy state to a lower-energy state, a(n) spectrum is produced. 18.

Assessment Chapter Test B - Ed W. Clark High School

Modern Chemistry Chapter 6 Chemical Bonding . Chemical Bond
•A link between atoms that results from the mutual attraction of their nuclei for electrons -Electrostatic attraction between proton and electron -Classified by the way the valence e- are distributed around nuclei of combined atoms .

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Chapter 12 Review 2 Multiple Choice Identify the letter of the choice that best completes the statement or answers the question. 1. binds the atoms together is called a(n) A mutual electrical attraction between the nuclei and valence electrons of different atoms that a. dipole. c. chemical bond. b. Lewis structure. d. London force. 2. a.