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electrons. Electrons In Atoms Vocabulary Review 5: Electrons in Atoms Vocabulary Review. The lowest-energy arrangement of electrons in a sh shell is obtained by putting electrons into depart

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orbitals of the s subshell before pairing electrons. Chapter Outline Review of Atomic Structure

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The modern description, primarily mathematical, of the behavior of electrons in atoms Atomic Orbital A mathematical expression describing the probability of finding an electron at various locations; Usually represented by the region of space around the nucleus where there is a high probability of finding an electron

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1) Electrons circulate around the nucleus 2) Electrons can jump from one level to the next but can NOT exist between levels (ladder) 3) Energy levels are NOT evenly spaced (escalator) 4) Energy lost or gained is NOT always the same 5) The higher the energy level, the easier for the electrons to leave (magnet)

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Number of Protons, Neutrons, and Electrons in an Atom. Date: 2020-2-12 | Size: 30.4Mb. If the atomic weight is 4.001, you can be confident the atom is helium, with 2 protons and 2 neutrons. An atomic weight closer to 5 is more troublesome. An anion carries a negative charge and has more electrons than protons.

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The electrons in an atom's outermost orbitals are called ---. Valence electrons. Each orbital may contain at most --- electrons. Two. All s orbitals are --- shaped. Spherically. A principal energy has --- energy sub levels.  $n^2$ . The maximum number of --- related to each principal energy level equals  $2n^2$ .

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Chemistry 2.2 Electrons in atoms Quiz Review. Negative particle found in an atom. Most of an atom. Positively charged center of an atom. Suggested that electrons orbit the atom's nucleus. Represents valence electrons. Units of frequency of a wave. Electromagnetic energy with the highest frequency / Highest energy.

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The center of an atom that is made up of protons and neutrons. Electron. A negatively charged subatomic particle that orbits around the nucleus of an atom. Electron cloud. The area in which

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Electrons orbit around the nucleus of an atom. Element. Pure substance made of a single type of atom. Atomic number.

### **Atoms, Isotopes & Ions Vocabulary Review Flashcards | Quizlet**

An atomic orbital may describe at most two electrons;  $1s^2 2s^2 2p^6$  Electrons enter orbitals of lowest energy first. the most stable arrangement of electrons around the nucleus of an atom Part D Questions and Problems Answer the following in the space provided. 22.

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