

## Chemistry Chapter 10 Assessment

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Section 10.1 Assessment page 324 7. Explain why chemists use the mole. ... 162 Chemistry: Matter and Change • Chapter 10 Solutions Manual CHAPTER 10 SOLUTIONS MANUAL 10. Explain how a mole is similar to a dozen. The mole is a unit for counting 6.02 10<sup>23</sup> representative particles. The dozen is used to

### The MoleThe Mole

Chemistry Chapter 10. diffusion. effusion. Graham's law of effusion. permeability. the spontaneous, uniform spreading and mixing of particles cau.... a process in which gas particles pass through a tiny opening i.... the gas law that states that the rate of effusion is inversely....

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1) The sum of all formal charges in a neutral molecule must be zero. 2) The sum of all formal charges in an ion must equal the charge of the ion. 3) Small (or zero) formal charges on individual atoms are better than large ones. 4) When formal charge cannot be avoided, negative formal charge should reside on the most electronegative atom.

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Solutions Manual Chemistry: Matter and Change • Chapter 11 211 CHAPTER 11 SOLUTIONS MANUAL Section 11.1 Assessment page 372 5. Compare the mass of the reactants and the mass of the products in a chemical reaction, and explain how these masses are related. The coefficients in the balanced equation indicate the molar relationship between each ...

## **StoichiometryStoichiometry**

CHEMISTRY: Chapter 10 Prep-Test Matching Match each item with the correct statement below. a. calorimeter d. temperature b. calorie e. specific heat c. joule f. heat \_\_\_\_ 1. quantity of heat needed to raise the temperature of 1 g of water by 1 C \_\_\_\_ 2. SI unit of energy \_\_\_\_ 3. energy transferred between 2 objects because of temperature difference

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