

Chem 1411 Chapter 2 Homework Answers

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Chem 1411 Chapter 2 Homework

CHEM 1411 Chapter 2 Homework Answers 1. Circle the molecules in the following list. H₂ HCl P₄ H₂O Fe S₈ CO₂ Ne 2. Circle the compounds in the following list. H₂ HCl P₄ H₂O Fe S₈ CO₂ Ne 3. Label the following as ionic or covalent: HCl - covalent P₄ - covalent H₂O - covalent NaCl - ionic Ca₃(PO₄)₂ - ionic CO₂ - covalent BaS - ionic 4.

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3 CHEM 1411. Chapter 2. Atoms, Molecules and Ions (Homework) 17. The elements in a column of the periodic table are known as A) metalloids. B) a period. C) noble gases. D) a group. E) nonmetals. 18. Write the formula for the acid formed from the fluoride anion, and then name the acid. 19. The mineral pyrolusite is a compound of manganese-55 and oxygen-16.

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7 CHEM 1411. Chapter 8. Molecular Geometry and Bonding Theories (Homework) W a. eleven sigma bonds and two pie bonds b. five sigma bonds and eleven pie bonds c. thirteen sigma bonds and one pi bond d. thirteen sigma bonds and two pie bonds e. eleven sigma bonds and five pie bonds Use the following Lewis structure for acetic acid to answer the following questions: ____ 36.

CHEM 1411. Chapter 9. Molecular Geometry and Bonding ...

Chem 1411 Homework Chapter 10 Question 1 1 out of 1 points A pressure of 0.500 atm is the same as a pressure of ____ of mm Hg. Selected Answer: 380 Correct Answer: 380 Question 2 1 out of 1 points A fixed amount of gas at 25.0 °C occupies a volume of 10.0 L when the pressure is 751 torr.

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1 CHEM 1411 Chapter 14 Homework 1. To increase the solubility in a liquid solvent, the solvent-solvent interactions should be ___ and the solute-solute interactions should be ___.

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